

PHYS 3446 – Lecture #10

Wednesday, Oct. 11, 2006

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1. Energy Deposition in Media

- Charged Particle Detection
- Ionization Process
- Photon Energy Loss



Announcements

- Colloquium today at 4pm in SH103
 - Dr. R. Arnowitt of Texas A&M
 - Title: Cosmology, SUSY and the LHC
 - Extra credit
- Quiz next Monday, Oct. 16
 - Covers CH4
- Reading assignment: CH5



Forces in Nature

- We have learned the discovery of two additional forces
 - Gravitational force: formulated through Newton's laws
 - Electro-magnetic force: formulated through Maxwell's equations
 - Strong nuclear force: Discovered through studies of nuclei and their structure
 - Weak force: Discovered and postulated through nuclear β -decay



Forewords

- Physics is an experimental science
 - Understand nature through experiments
- In nuclear and particle physics, experiments are performed through scattering of particles
- In order for a particle to be detected:
 - Must leave a trace of its presence → deposit energy

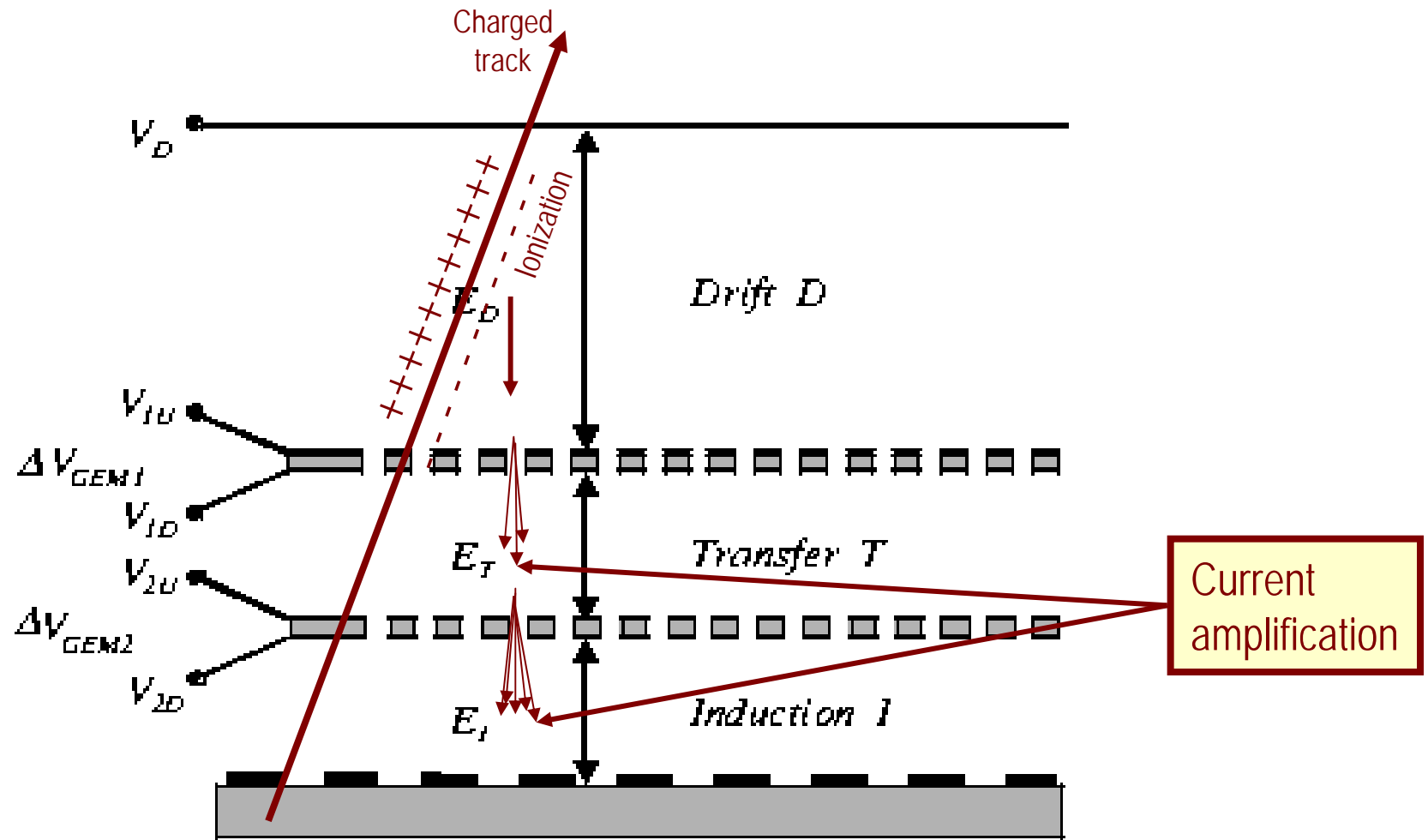


Forewords

- The most ideal detector should
 - Detect particle without affecting them
- Realistic detectors
 - Use electromagnetic interactions of particles with matter
 - Ionization of matter by energetic, charged particles
 - Ionization electrons can then be accelerated within an electric field to produce detectable electric current
 - Sometime catastrophic nuclear collisions but rare
 - Particles like neutrinos which do not interact through EM and have low cross sections, need special methods to handle



How does a charged particle get detected?



C. GEMs

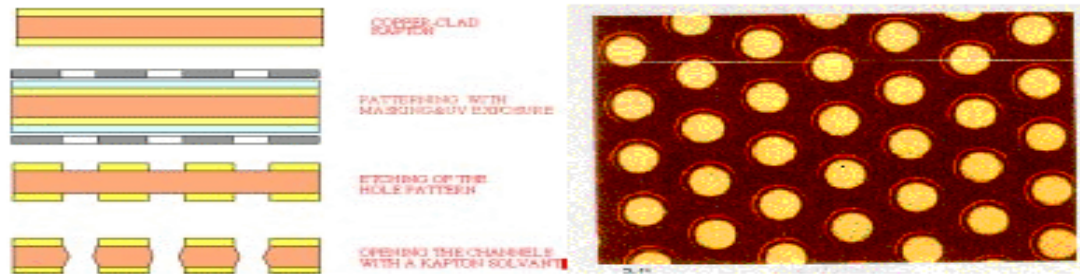


Fig. 14(a) Chemical etching Process of a GEM (b) A GEM foil

A new concept of gas amplification was introduced in 1996 by Sauli: the Gas Electron Multiplier (GEM) [27] manufactured by using standard printed circuit wet etching techniques, schematically shown in Fig. 14(a). Comprising a thin ($\sim 50 \mu\text{m}$) Kapton foil, double sided clad with Copper, holes are perforated through (fig. 15b). The two surfaces are maintained at a potential gradient, thus providing the necessary field for electron amplification, as shown in Fig. 15(a), and an avalanche of electrons as in Fig. 15(b).

Large amplification

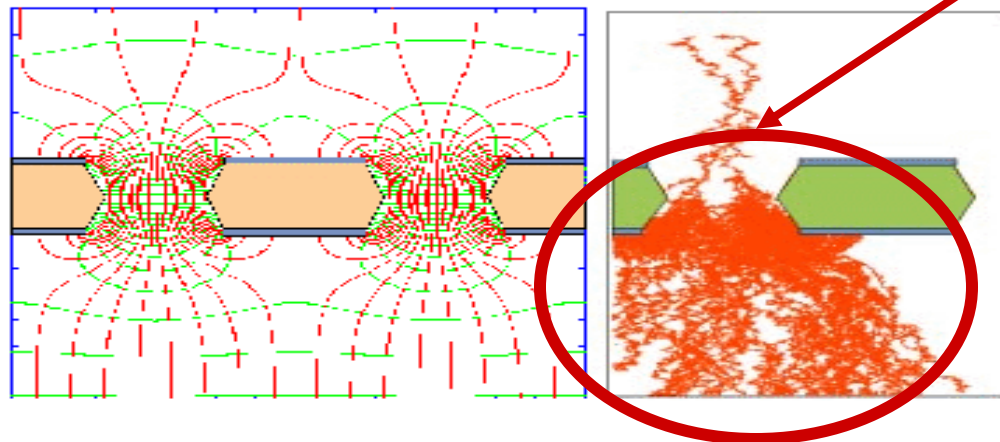
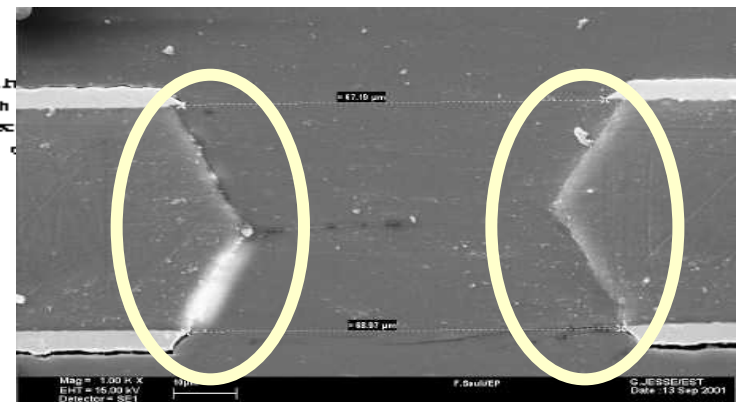
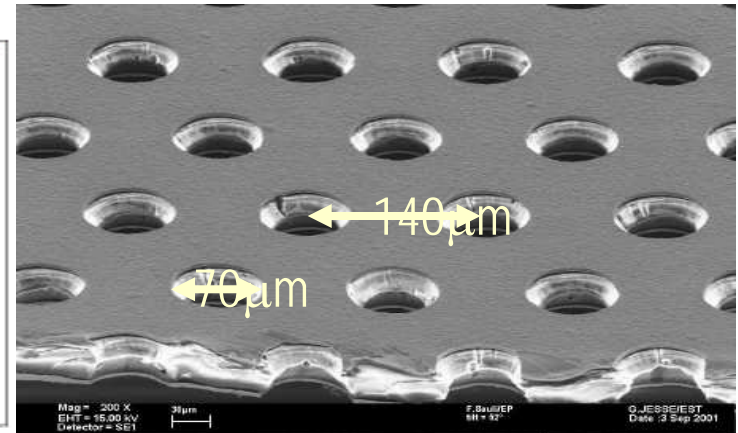


Fig. 15(a) Electric Field and (b) an avalanche across a GEM channel

Coupled with a drift electrode above and a readout electrode below, it acts as a highly performant micro-patterned detector. The essential and advantageous feature of this detector is that amplification, detection are decoupled, and the readout is at zero potential. Permitting charge transfer to a second amplification device, this opens up the possibility of using a GEM in tandem with an MSGC or second GEM.



CERN-open-2000-344, A. Sharma

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Charged Particle Detection

- What do you think is the primary interaction when a charged particle is traversing through a medium?
 - Interactions with the atomic electrons in the medium
- If the energy of the charged particle is sufficiently high
 - It deposits its energy (or loses its energy in the matter) by ionizing the atoms in the path electrons
 - Or by exciting atoms or molecules to higher states photons
 - What are the differences between the above two methods?
 - The outcomes are either electrons or photons
- If the charged particle is massive, its interactions with atomic electrons will not affect the particles trajectory
- Sometimes, the particle undergoes a more catastrophic nuclear collisions



Ionization Process

- Ionization properties can be described by the stopping power variable, $S(T)$
 - Definition: amount of kinetic energy lost by any incident object per unit length of the path traversed in the medium
 - Referred as ionization energy loss or energy loss

$$S(T) = -\frac{dT}{dx} = n_{ion} \bar{I}$$

Why negative sign?

The particle's energy decreases.

- T : Kinetic energy of the incident particle
- n_{ion} : Number of electron-ion pair formed per unit path length
- \bar{I} : The average energy needed to ionize an atom in the medium; for large atomic numbers $\sim 10Z$ eV.

Ionization Process

- What do you think the stopping power of the given medium depends on?
 - Energy of the incident particle
 - Depends very little for relativistic particles
 - Electric charge of the incident particle
- Since ionization is an EM process, easily calculable
 - Bethe-Bloch formula for relativistic particle

$$S(T) = \frac{4\pi (ze)^2 e^2 nZ}{m\beta^2 c^2} \left[\ln \left(\frac{2mc^2 \gamma^2 \beta^2}{\bar{I}} \right) - \beta^2 \right]$$

- z: Incident particle atomic number
- Z: medium atomic number
- n: number of atoms in unit volume ($=\rho A_0/A$)
- m: mass of the medium



Ionization Process

- In natural α -decay, the formula becomes

$$S(T) = \frac{4\pi (ze)^2 e^2 nZ}{m\beta^2 c^2} \left[\ln \left(\frac{2mc^2 \cancel{\gamma^2} \beta^2}{\bar{I}} \right) - \cancel{\beta^2} \right] \approx \frac{4\pi (ze)^2 e^2 nZ}{m\beta^2 c^2} \ln \left(\frac{2mc^2 \beta^2}{\bar{I}} \right)$$

1 0

- Due to its low kinetic energy (a few MeV) and large mass, relativistic corrections can be ignored
- For energetic particles in accelerator experiments or beta emissions, the relativistic corrections are substantial
- Bethe-Bloch formula can be used in many media, various incident particles over a wide range of energies



Ionization Process

- Why does the interaction with atomic electrons dominate the energy loss of the incident particle?
 - Interactions with heavy nucleus causes large change of direction of the momentum but little momentum transfer
 - Does not necessarily require large loss of kinetic energy
 - While momentum transfer to electrons would require large kinetic energy loss
 - Typical momentum transfer to electrons is 0.1MeV/c and requires 10KeV of kinetic energy loss
 - The same amount of momentum transfer to a gold nucleus would require less than 0.1eV of energy loss
- Thus Bethe-Bloch formula is inversely proportional to the mass of the medium

$$S(T) = \frac{4\pi (ze)^2 e^2 nZ}{m\beta^2 c^2} \left[\ln \left(\frac{2mc^2 \gamma^2 \beta^2}{\bar{I}} \right) - \beta^2 \right]$$

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Ionization Process

- At low particle velocities, ionization loss is sensitive to particle energy. How do you see this?

$$S(T) = \frac{4\pi (ze)^2 e^2 nZ}{m\beta^2 c^2} \left[\ln \left(\frac{2mc^2 \gamma^2 \beta^2}{\bar{I}} \right) - \beta^2 \right]$$

– Stopping power decreases as v increases!!

- This shows that the particles of different rest mass (M) but the same momentum (p) can be distinguished due to their different energy loss rate

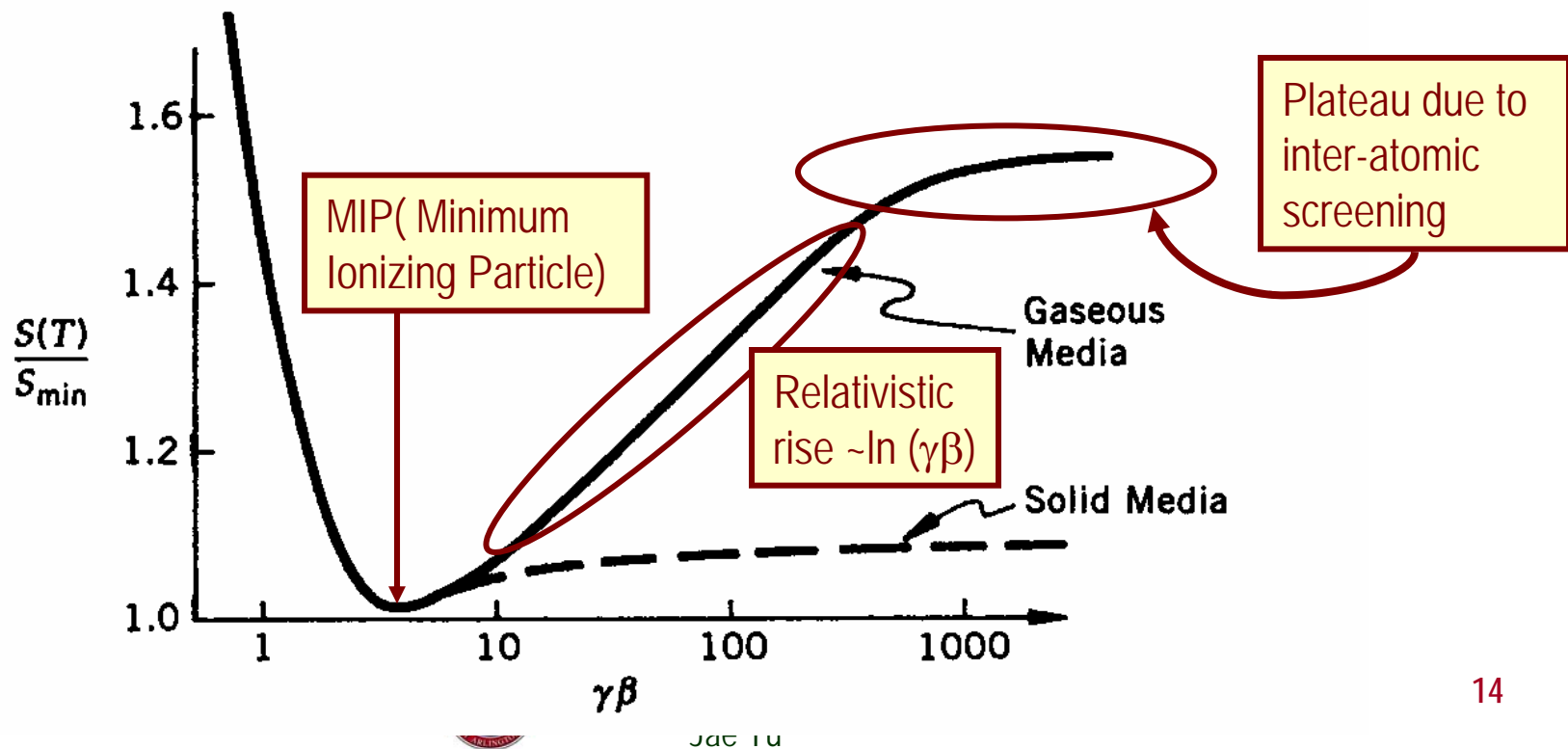
$$S(T) \propto \frac{1}{v^2} = \frac{1}{(\beta c)^2} = \frac{M^2 \gamma^2}{(M \gamma \beta c)^2} = \frac{M^2 \gamma^2}{p^2}$$

- At low velocities ($\gamma \sim 1$), particles can be distinguished



Properties of Ionization Process

- Stopping power decreases with increasing particle velocity independent of incident particle mass
 - Minimum occurs when $\gamma\beta \sim 3$
 - Particle is minimum ionizing when $v \sim 0.96c$
 - For massive particles the minimum occurs at higher momenta
 - This is followed by a $\ln(\gamma\beta)$ relativistic rise by Beth-Bloch formula
 - Energy loss plateaus at high $\gamma\beta$ due to long range inter-atomic screening effect which is ignored in Beth-Bloch



Ionization Process

- At very high energies
 - Relativistic rise becomes an energy independent constant rate
 - Cannot be used to distinguish particle-types purely using ionization
 - Except for gaseous media, the stopping power at high energies can be approximated by the value at $\gamma\beta \sim 3$.
- At low energies, the stopping power expectation becomes unphysical
 - Ionization loss is very small when the velocity is very small
 - Detailed atomic structure becomes important



Ranges of Ionization Process

- Once the stopping power is known, we can compute the expected range of any particle in the medium
 - The distance the incident particle can travel in the medium before its kinetic energy runs out

$$R = \int_0^R dx = \int_T^0 \frac{dx}{dT} dT = \int_0^T \frac{dT}{S(T)}$$

- At low E, two particles with same KE but different mass can have very different ranges
 - This is why α and β radiations have quite different requirements to stop



Units of Energy Loss and Range

- What would be the sensible unit for energy loss?
 - MeV/cm
 - Equivalent thickness of g/cm²: MeV/(g/cm²)
- Range is expressed in
 - cm or g/cm²
- Minimum value of S(T) for z=1 at $\gamma\beta=3$ is

$$S(T)_{\min} \approx -\frac{4\pi e^4 A_0 (\rho Z/A)}{m\beta^2 c^2} \ln\left(\frac{2mc^2 \gamma^2 \beta^2}{\bar{I}}\right) \approx 5.2 \times 10^{-7} (13.7 - \ln Z) \rho Z/A \text{ erg/cm}$$

- Using $\langle Z \rangle = 20$ we can approximate

$$S(T)_{\min} \approx 3.5 \frac{Z}{A} \text{ MeV/(g/cm}^2\text{)}$$



Straggling, Multiple Scattering and Statistical process

- Phenomenological calculations can describe average behavior but large fluctuations are observed in an event-by-event bases
 - This is due to the statistical nature of scattering process
 - Finite dispersion of energy deposit or scattering angular distributions is measured
- Statistical effect of angular deviation experienced in Rutherford scattering off atomic electrons in the medium
 - Consecutive collisions add up in a random fashion and provide net deflection of any incident particles from its original path
 - Called “Multiple Coulomb Scattering” → Increases as a function of path length

$$\theta_{rms} \approx \frac{20 MeV}{\beta pc} z \sqrt{\frac{L}{X_0}}$$

- z: charge of the incident particle, L: material thickness, X_0 : radiation length of the medium



Energy Loss Through Bremsstrahlung

- Energy loss of incident electrons
 - Bethe-Bloch formula works well (up to above 1MeV for electrons)
 - But due to the small mass, electron's energy loss gets complicated
 - Relativistic corrections take large effect even down to a few keV level
 - Electron projectiles can transfer large fractions of energies to the atomic electrons they collide
 - Produce δ -rays or knock-on electrons → Which have the same properties as the incident electrons
 - Electrons suffer large acceleration as a result of interaction with electric field by nucleus. What do these do?
 - Causes electrons to radiate or emit photons
 - Bremsstrahlung → An important mechanism of relativistic electron energy loss

