

# PHYS 3313 – Section 001

## Lecture #17

*Monday, April 6, 2015*

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- Normalization and Probability
- Time Independent Schrodinger Equation



# Announcements

- Research paper template has been placed onto the class web page link to research
- Homework #4
  - End of chapter problems on CH5: 8, 10, 16, 24, 26, 36 and 47
  - Due Monday, Apr. 13
- Quiz #4 at the beginning of the class Monday, Apr. 13
  - Covers CH 5.4 through what we finish this Wednesday
- Colloquium this Wed.



# Reminder: Special Project #4

- Prove that the wave function  $\Psi = A[\sin(kx - \omega t) + i\cos(kx - \omega t)]$  is a good solution for the time-dependent Schrödinger wave equation. Do NOT use the exponential expression of the wave function. (10 points)
- Determine whether or not the wave function  $\Psi = Ae^{-\alpha|x|}$  satisfy the time-dependent Schrödinger wave equation. (10 points)
- Due for this special project is Wednesday, Apr. 8.
- You MUST have your own answers!



# Normalization and Probability

- The probability  $P(x) dx$  of a particle being between  $x$  and  $x + dx$  was given by the equation

$$P(x)dx = \Psi^*(x,t)\Psi(x,t)dx$$

- Here  $\Psi^*$  denotes the complex conjugate of  $\Psi$
- The probability of the particle being between  $x_1$  and  $x_2$  is given by

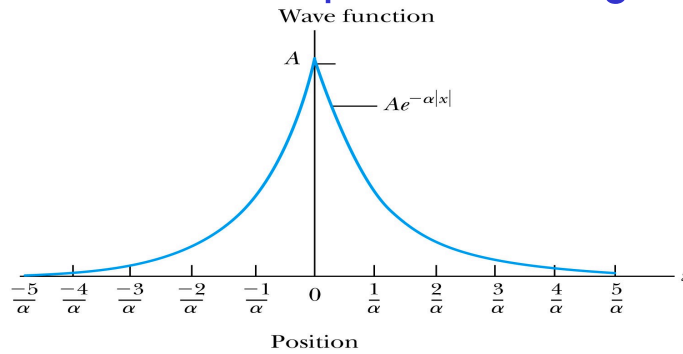
$$P = \int_{x_1}^{x_2} \Psi^* \Psi dx$$

- The wave function must also be normalized so that the probability of the particle being somewhere on the  $x$  axis is 1.

$$\int_{-\infty}^{+\infty} \Psi^*(x,t)\Psi(x,t)dx = 1$$

# Ex 6.4: Normalization

Consider a wave packet formed by using the wave function that  $Ae^{-\alpha|x|}$ , where  $A$  is a constant to be determined by normalization. Normalize this wave function and find the probabilities of the particle being between 0 and  $1/\alpha$ , and between  $1/\alpha$  and  $2/\alpha$ .



$$\Psi = Ae^{-\alpha|x|}$$

Probability density

$$\int_{-\infty}^{+\infty} \Psi^* \Psi dx = \int_{-\infty}^{+\infty} (Ae^{-\alpha|x|})^* (Ae^{-\alpha|x|}) dx = \int_{-\infty}^{+\infty} (A^* e^{-\alpha|x|}) (Ae^{-\alpha|x|}) dx =$$

$$= \int_{-\infty}^{+\infty} A^2 e^{-2\alpha|x|} dx = 2 \int_0^{+\infty} A^2 e^{-2\alpha x} dx = \left. \frac{2A^2}{-2\alpha} e^{-2\alpha x} \right|_0^{+\infty} = 0 + \frac{A^2}{\alpha} = 1$$

$$A = \sqrt{\alpha}$$

Normalized Wave Function

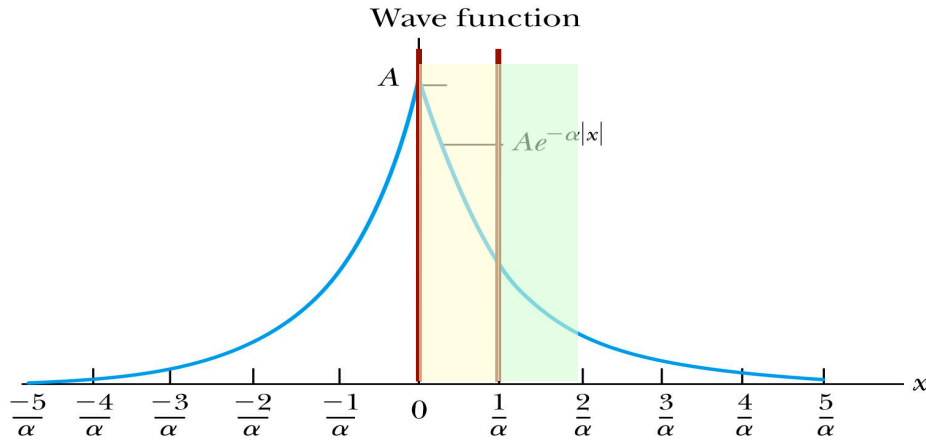
$$\Psi = \sqrt{\alpha} e^{-\alpha|x|}$$

## Ex 6.4: Normalization, cont'd

Using the wave function, we can compute the probability for a particle to be with 0 to  $1/\alpha$  and  $1/\alpha$  to  $2/\alpha$ .

$$\Psi = \sqrt{\alpha} e^{-\alpha|x|}$$

For 0 to  $1/\alpha$ :



$$P = \int_0^{1/\alpha} \Psi^* \Psi dx = \int_0^{1/\alpha} \alpha e^{-2\alpha x} dx = \frac{\alpha}{-2\alpha} e^{-2\alpha x} \Big|_0^{1/\alpha} = -\frac{1}{2} (e^{-2} - 1) \approx 0.432$$

For  $1/\alpha$  to  $2/\alpha$ :

$$P = \int_{1/\alpha}^{2/\alpha} \Psi^* \Psi dx = \int_{1/\alpha}^{2/\alpha} \alpha e^{-2\alpha x} dx = \frac{\alpha}{-2\alpha} e^{-2\alpha x} \Big|_{1/\alpha}^{2/\alpha} = -\frac{1}{2} (e^{-4} - e^{-2}) \approx 0.059$$

How about  $2/\alpha$  to  $\infty$ ?

# Properties of Valid Wave Functions

## Boundary conditions

- 1) To avoid infinite probabilities, the wave function must be finite everywhere.
- 2) To avoid multiple values of the probability, the wave function must be single valued.
- 3) For finite potentials, the wave function and its derivatives must be continuous. This is required because the second-order derivative term in the wave equation must be single valued. (There are exceptions to this rule when  $V$  is infinite.)
- 4) In order to normalize the wave functions, they must approach zero as  $x$  approaches infinity.

**Solutions that do not satisfy these properties do not generally correspond to physically realizable circumstances.**



# Time-Independent Schrödinger Wave Equation

- The potential in many cases will not depend explicitly on time.
- The dependence on time and position can then be separated in the Schrödinger wave equation. Let,  $\Psi(x,t) = \psi(x)f(t)$

which yields: 
$$i\hbar\psi(x)\frac{\partial f(t)}{\partial t} = -\frac{\hbar^2 f(t)}{2m}\frac{\partial^2 \psi(x)}{\partial x^2} + V(x)\psi(x)f(t)$$

Now divide by the wave function: 
$$i\hbar\frac{1}{f(t)}\frac{\partial f(t)}{\partial t} = -\frac{\hbar^2 f(t)}{2m}\frac{1}{\psi(x)}\frac{\partial^2 \psi(x)}{\partial x^2} + V(x)$$

- The *left side of this last equation* depends only on time, and *the right side* depends only on spatial coordinates. Hence each side must be equal to a constant. The time dependent side is

$$i\hbar\frac{1}{f}\frac{df}{dt} = B$$





# Time-Independent Schrödinger Wave Equation(con't)

- We integrate both sides and find:  $i\hbar \int \frac{df}{f} = \int B dt \Rightarrow i\hbar \ln f = Bt + C$

where  $C$  is an integration constant that we may choose to be 0.

Therefore

$$\ln f = \frac{Bt}{i\hbar}$$

This determines  $f$  to be  $f(t) = e^{Bt/i\hbar} = e^{-iBt/\hbar}$ . Comparing this to the time dependent portion of the free particle wave function  $e^{-i\omega t} = e^{-iBt/\hbar}$

$$\Rightarrow B = \hbar\omega = E \quad \Rightarrow \quad i\hbar \frac{1}{f(t)} \frac{\partial f(t)}{\partial t} = E$$

- This is known as the **time-independent Schrödinger wave equation**, and it is a fundamental equation in quantum mechanics.

$$-\frac{\hbar^2}{2m} \frac{d^2\psi(x)}{dx^2} + V(x)\psi(x) = E\psi(x)$$

# Stationary State

- Recalling the separation of variables:  $\Psi(x,t) = \psi(x)f(t)$   
and with  $f(t) = e^{-i\omega t}$  the wave function can be written as:  $\Psi(x,t) = \psi(x)e^{-i\omega t}$

- The probability density becomes:

$$\Psi^* \Psi = \psi^2(x) \left( e^{i\omega t} e^{-i\omega t} \right) = \psi^2(x)$$

- The probability distributions are constant in time.  
This is a standing wave phenomena that is called the stationary state.



# Comparison of Classical and Quantum Mechanics

- Newton's second law and Schrödinger's wave equation are both differential equations.
- Newton's second law can be derived from the Schrödinger wave equation, so the latter is the more fundamental.
- Classical mechanics only appears to be more precise because it deals with macroscopic phenomena. The underlying uncertainties in macroscopic measurements are just too small to be significant.

