PHYS 1442 – Section 001 Lecture #2

Tuesday, June 4, 2013 Dr. **Jae**hoon **Yu**

- Chapter 16
 - Insulators and Conductors & Induced Charge
 - Coulomb's Law
 - Vector Operations Recap
 - The Electric Field & Field Lines
 - Electric Fields and Conductors
 - Motion of a Charged Particle in an Electric Field
 - Gauss' Law



Announcements

- 22 of you have registered in the homework system.
 - Fantastic job!!
 - You need my enrollment approval... So move quickly...
 - Remember, the deadline for the first homework is 11pm this Thursday.
- Reading assignments: CH16 10, 16 11, 16 12
- Quiz at the beginning of the class this Thursday, June 6
 Appendix A1 A8 and CH16



Extra Credit Special Project #1

- Compare the Coulomb force to the Gravitational force in the following cases by expressing Coulomb force $(F_{\rm C})$ in terms of the gravitational force (F_{G})
 - Between two protons separated by 1m
 - Between two protons separated by an arbitrary distance R
 - Between two electrons separated by 1m
 - Between two electrons separated by an arbitrary distance R
- Five points each, totaling 20 points
- BE SURE to show all the details of your work, including all formulae, and properly referring them
- Please staple them before the submission
- Due at the beginning of the class Monday, June 10



Special Project #2 – Angels & Demons

- Compute the total possible energy released from an annihilation of x-grams of anti-matter and the same quantity of matter, where x is the last two digits of your SS#. (20 points)
 - Use the famous Einstein's formula for mass-energy equivalence
- Compute the power output of this annihilation when the energy is released in x ns, where x is again the last two digits of your SS#. (10 points)
- Compute how many cups of gasoline (8MJ) this energy corresponds to. (5 points)
- Compute how many months of world electricity usage (3.6GJ/mo) this energy corresponds to. (5 points)
- Due by the beginning of the class Thursday, June 13.



Insulators and Conductors

- Let's imagine two metal balls of which one is charged
- What will happen if they are connected by
 - A metallic object?
 - Some charge is transferred.
 - These objects are called conductors of electricity.
 - A wooden object?
 - No charge is transferred
 - These objects are called nonconductors or insulators.
- Metals are generally good conductors whereas most other materials are insulators.
 - There are third kind of materials called, semi-conductors, like silicon or germanium \rightarrow conduct only in certain conditions
- Atomically, conductors have loosely bound electrons while insulators have them tightly bound!

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Metal



Charged Neutral



+ +

Induced Charge

- When a positively charged metal object is brought close to an uncharged metal object
 - If two objects touch each other, the free electrons in the neutral ones are attracted to the positively charged object and some will pass over to it, leaving the neutral object positively charged. \rightarrow Charging by conduction
 - If the objects get close, the free electrons in the neutral ones still move within the metal toward the charged object leaving the opposite of the object positively charged.
 - The charges have been "induced" in the opposite ends of the object Tuesday, June 4, 2013





Induced Charge

- We can induce a net charge on a metal object by connecting a wire to the ground.
 - The object is "grounded" or "earthed".
- Since the Earth is so large and conducts, it can give or accept charge.
 - The Earth acts as a reservoir of the electric charge.
- If the negative charge is brought close to a neutral metal
 - The positive charges will be induced toward the negatively charged metal.
 - The negative charges in the neutral metal will be gathered on the opposite side, transferring through the wire to the Earth.
 - If the wire is cut, the metal bar has net positive charge.
- An <u>electroscope</u> is a device that can be used for detecting charge and signs.
 - Previously charged device can determine the sign of unknown charge

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ground (a) ++ (b)+ + (c) Insulator Metal Gold leaves Glass

Coulomb's Law

- Electric charges exert force to each other. What factors affect the magnitude of this force?
 - Any guesses?
- Charles Coulomb figured this out in 1780's.
- Coulomb found that the electrical force is
 - Proportional to the multiplication of the two charges
 - If one of the charges doubles, the force doubles.
 - If both the charges double, the force quadruples.
 - Inversely proportional to the square of the distances between them.
 - If the distance between the two charges double, the force reduces by a factor of 4!
 - Electric charge is a fundamental property of matter, just like mass.
- How would you put the above into a formula?





- Is Coulomb force a scalar quantity or a vector quantity? Unit?
 - A vector quantity. The unit is Newtons (N)!
- The direction of electric (Coulomb) force is always along the line joining the two objects.
 - If the two charges are the same: forces are directed away from each other.
 - If the two charges are opposite: forces are directed toward each other.
- Coulomb force is precise to 1 part in 10¹⁶.
- Unit of charge is called Coulomb, C, in SI.
- The value of the proportionality constant, $k_{\rm s}$ in SI unit is $k = 8.988 \times 10^9 \text{ N} \cdot \text{m}^2/\text{C}^2$
- Thus, 1C is the charge that gives F~9x10⁹N of force when placed 1m apart from each other.



- Does the electric force look similar to another force? What is it?
 - Gravitational Force
- What are the sources of the forces?
 - Electric Force: Electric charges, fundamental properties of matter
 - Gravitational Force: Masses, fundamental properties of matter
- What else is similar?
 - Inversely proportional to the square of the distance between the sources of the force → What is this kind law called?
 - Inverse Square Law
- What is the biggest difference?
 - Gravitational force is always attractive.
 - Electric force depends on the type of the two charges.



The Elementary Charge and Permittivity

- Elementary charge, the smallest charge, is that of an electron: $e = 1.602 \times 10^{-19} C$
 - Since an electron is a negatively charged particle, its charge is -e.
- Object cannot gain or lose fraction of an electron.
 - Electric charge is quantized.
 - It changes always in integer multiples of *e*.
- The proportionality constant k is often written in terms of another constant, ε_0 , the permittivity* of free space. They are related $k = 1/4\pi\varepsilon_0$ and $\varepsilon_0 = 1/4\pi k = 8.85 \times 10^{-12} C^2/N \cdot m^2$.
- Thus the electric force can be written: $F = \frac{1}{4\pi\varepsilon_0} \frac{Q_1 Q_2}{r^2}$
- Note that this force is for "point" charges at rest.

*Mirriam-Webster, Permittivity: The ability of a material to store electric potential energy under the influence of an electric field



Example 16 – 1

• Electric force on electron by proton. Determine the magnitude of the electric force on the electron of a hydrogen atom exerted by the single proton (Q_2 =+e) that is its nucleus. Assume the electron "orbits" the proton at its average distance of r=0.53x10⁻¹⁰m.



Using Coulomb's law
$$F = \frac{1}{4\pi\varepsilon_0} \frac{Q_1 Q_2}{r^2} = k \frac{Q_1 Q_2}{r^2}$$

Each charge is $Q_1 = -e = -1.602 \times 10^{-19} C$ and $Q_2 = +e = 1.602 \times 10^{-19} C$

So the magnitude of the force is

$$F = \left| k \frac{Q_1 Q_2}{r^2} \right| = 9.0 \times 10^9 \, N \cdot m^2 / C^2 \frac{\left(1.6 \times 10^{-19} \, C \right) \left(1.6 \times 10^{-19} \, C \right)}{\left(0.53 \times 10^{-10} \, m \right)^2}$$
$$= 8.2 \times 10^{-8} \, N$$

Which direction?

Toward each other...



Example 16 - 2

• Which charge exerts a greater force? Two positive point charges, $Q_1 = 50\mu$ C and $Q_2 = 1\mu$ C, are separated by a distance L. Which is larger in magnitude, the force that Q_1 exerts on Q_2 or the force that Q_2 exerts on Q_1 ?

What is the force that Q_1 exerts on Q_2 ?

$$F_{12} = k \frac{Q_1 Q_2}{L^2}$$

What is the force that Q_2 exerts on Q_1 ?

$$F_{21} = k \frac{Q_2 Q_1}{L^2}$$

Therefore the magnitudes of the two forces are identical!!

Well then what is different? The direction.

Which direction?

What is this law?

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Opposite to each other!

Newton's third law, the law of action and reaction!!



Vector Additions and Subtractions

- Addition:
 - Triangular Method: One can add vectors by connecting the head of one vector to the tail of the other (head-to-tail)
 - Parallelogram method: Connect the tails of the two vectors and extend
 - Addition is commutative: Changing order of operation does not affect the results A +B=B+A, A+B+C+D+E=E+C+A+B+D



- Subtraction:
 - The same as adding a negative vector: **A B** = **A** + (-**B**)



Since subtraction is the equivalent to adding a negative vector, subtraction is also commutative!!!

• Multiplication by a scalar is increasing the magnitude A, B=2A Tuesda B=2AB=2A B=2ADr. Jaehoon Yu 14

Example for Vector Addition

A force of 20.0N applies to north while another force of 35.0N applies in the direction 60.0° west of north. Find the magnitude and direction of resultant force.



Components and Unit Vectors

Coordinate systems are useful in expressing vectors in their components



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Unit Vectors

- Unit vectors are the ones that tells us the directions of the components
- Dimensionless
- Magnitudes are exactly 1
- Unit vectors are usually expressed in i, j, k or

$$\vec{i}, \vec{j}, \vec{k}$$

So the vector **F** can be re-written as

$$\vec{F} = F_x \vec{i} + F_y \vec{j} = \left| \vec{F} \right| \cos \theta \vec{i} + \left| \vec{F} \right| \sin \theta \vec{j}$$



Examples of Vector Operations

Find the resultant force which is the sum of **F1**=(2.0**i**+2.0**j**)N and **F2**=(2.0**i**-4.0**j**)N.

$$\vec{F}_{3} = \vec{F}_{1} + \vec{F}_{2} = (2.0\vec{i} + 2.0\vec{j}) + (2.0\vec{i} - 4.0\vec{j})$$
$$= (2.0 + 2.0)\vec{i} + (2.0 - 4.0)\vec{j} = 4.0\vec{i} - 2.0\vec{j}(N)$$
$$\left|\vec{F}_{3}\right| = \sqrt{(4.0)^{2} + (-2.0)^{2}}$$
$$\theta = \tan^{-1}\frac{F_{3y}}{F_{3x}} = \tan^{-1}\frac{-2.0}{4.0} = -27^{\circ}$$

Find the resultant force of the sum of three forces: $F_1 = (15i+30j+12k)N$, $F_2 = (23i+14j-5.0k)N$, and $F_3 = (-13i+15j)N$.

$$\vec{F} = \vec{F}_1 + \vec{F}_2 + \vec{F}_3 = (15\vec{i} + 30\vec{j} + 12\vec{k}) + (23\vec{i} + 14\vec{j} - 5.0\vec{k}) + (-13\vec{i} + 15\vec{j})$$

= $(15 + 23 - 13)\vec{i} + (30 + 14 + 15)\vec{j} + (12 - 5.0)\vec{k} = 25\vec{i} + 59\vec{j} + 7.0\vec{k}(N)$

gnitude
$$\left| \vec{D} \right| = \sqrt{(25)^2 + (59)^2 + (7.0)^2} = 65(N)$$

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Example 16 – 3

• Three charges in a line. Three charged particles are arranged in a line as shown in the figure. Calculate the net electrostatic force on particle 3 (the -4µC on the right) due to other two charges. What is the force that Q₁ exerts on Q₃? $F_{13x} = k \frac{Q_1Q_3}{L^2} = \frac{(9.0 \times 10^9 \ N \cdot m^2/C^2)(-4.0 \times 10^{-6} \ C)(-8.0 \times 10^{-6} \ C)}{(0.5m)^2} = 1.2N$ What is the force that Q₂ exerts on Q₃? $F_{23x} = k \frac{Q_2Q_3}{L^2} = \frac{(9.0 \times 10^9 \ N \cdot m^2/C^2)(-4.0 \times 10^{-6} \ C)(3.0 \times 10^{-6} \ C)}{(0.2m)^2} = -2.7N$

Using the vector sum of the two forces

$$F_x = F_{13x} + F_{23x} = 1.2 + (-2.7) = -1.5(N) \qquad F_y = 0(N)$$

$$\vec{F} = -1.5\vec{i}(N)$$

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—0.30 m→+•0.20 m→