## PHYS 1443 – Section 003 Lecture #1

Monday, Aug. 25, 2003 Dr. **Jae**hoon Yu

- 1. Who am I?
- 2. How is this class organized?
- 3. What is Physics?
- 4. What do we want from this class?
- 5. Brief history of physics
- 6. Chapter one
  - Uncertainties and Significant Figures
  - Standards and units
  - Estimates
  - Dimensional Analysis

Today's homework is homework #1, due 1pm, next Wednesday!!

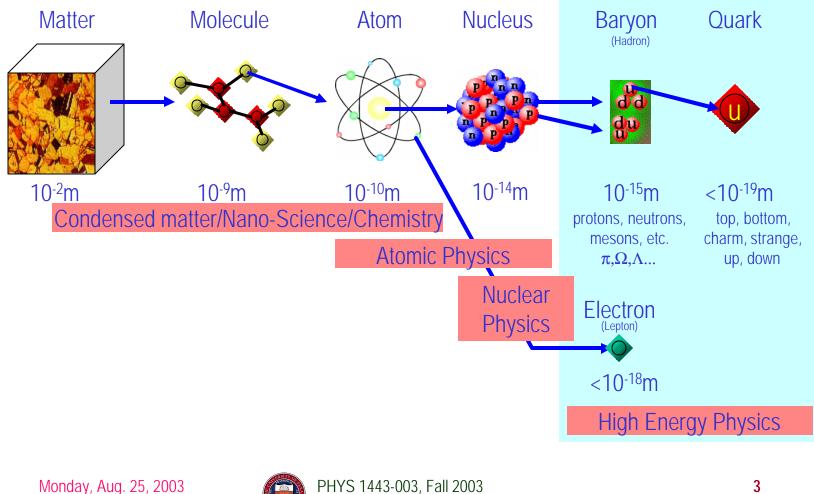


## Who am I?

- Name: Dr. Jaehoon Yu (You can call me Dr. Yu)
- Office: Rm 242A, Science Hall
- Extension: x2814, E-mail: *jaehoonyu@uta.edu*
- My profession: High Energy Physics (HEP)
  - Collide particles (protons on anti-protons or electrons on anti-electrons, positrons) at the energies equivalent to 10,000 Trillion degrees
  - To understand
    - Fundamental constituents of matter
    - Interactions or forces between the constituents
    - Creation of Universe (Big Bang Theory)
  - A pure scientific research activity
    - Direct use of the fundamental laws we find may take longer than we want but
    - Indirect product of research contribute to every day lives; eg. WWW



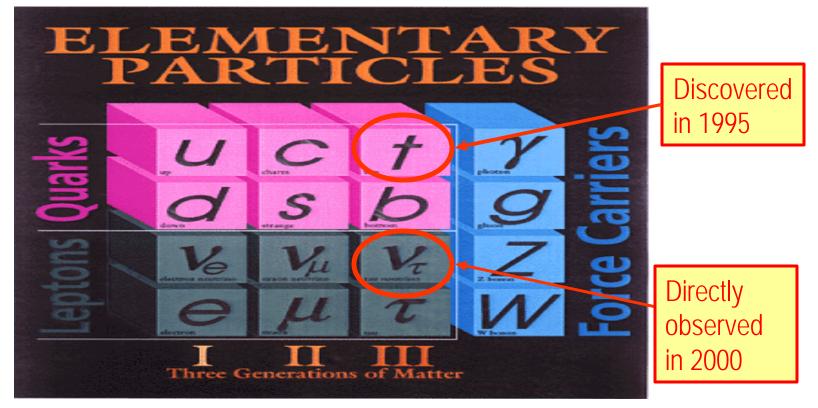
#### **Structure of Matter**



Dr. Jaehoon Yu

### The Standard Model

#### • Assumes the following fundamental structure:



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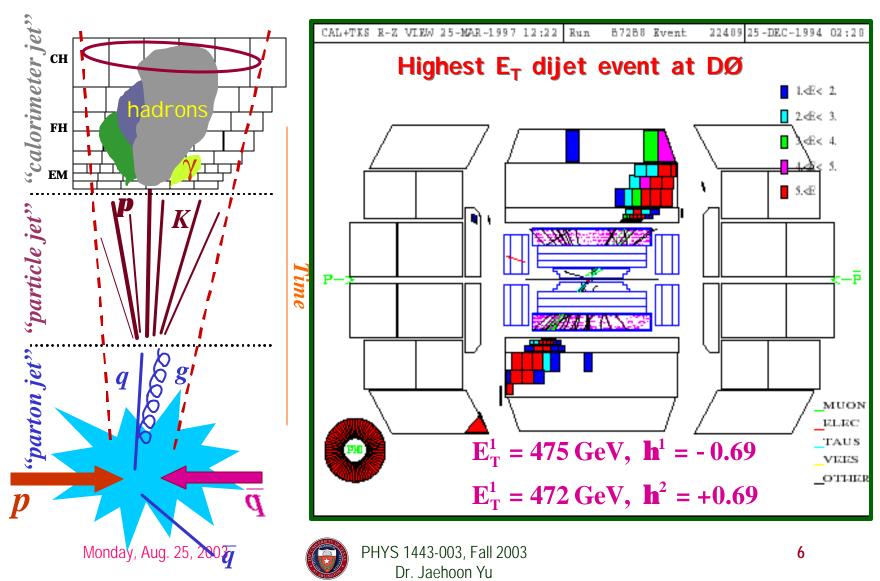
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### Fermilab Tevatron Accelerator

- World's Highest Energy proton-anti-proton collider
  - E<sub>cm</sub>=1.96 TeV (=6.3x10<sup>-7</sup>J/p→ 1.3MJoule)



#### How does an Event Look in a Collider Detector?



#### Information & Communication Source

- My web page: <u>http://www-hep.uta.edu/~yu/</u>
  - Contact information & Class Schedule
  - Syllabus
  - Holidays and Exam days
  - Evaluation Policy
  - Class Style & homework
  - Other information
- Primary communication tool is e-mail: Register for <u>PHYS1443-003-FALL03 e-mail distribution list</u> as soon possible
- Office Hours: 2:30 3:30pm, Mondays and Wednesdays



#### Primary Web Page

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| Course Specification for 1443-003, Fall 2003   |  |           |  |  |
| Class Schedule   | 1:00 - 2:20pm<br>Mondays & Wednesdays<br>Room 103, Science Hall  |           |  |  |
| Instructor   | Dr. Jaehoon Yu   |           |  |  |
| Office   | Room 242A, Science Hall<br>Phone: (817) 272 - 2814<br>Secretary: (817) 272 - 2811<br>Fermilab Office: (630) 840 - 8308 |           |  |  |
| Office Hours   | 2:30pm - 3:30pm<br>Mondays and Wednesdays  |           |  |  |
| Prerequisites  | MATH 1426 or concurrent enrollment.<br>You must enroll in a relevant <u>lab</u> section, unless exempt                 |           |  |  |
| Textbook   | Physics for Scientists and Engineers, 3 <sup>rd</sup> Edition<br>Douglas C. Giancoli<br>Prentice Hall                  |           |  |  |
| Syllabus     Lecture Notes     Home Work Assignments     Link to Physics Labs     Term Exam Problems and Solutions |  |           |  |  |
| Class Style & Communication methods     Holidays & Exam Days     Evaluation Policy     Contact Information         |  |           |  |  |
| Other Information  |  |           |  |  |
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| Syllabu   | s for1443-003, Fall 2003  |
| Period  | Chapters  |
| Weeks of Aug. 25 and Sept. 1, 2003                      | Chapters 1 - 3:<br>Estmiate, measurements, One and two dimensional motion<br>(No class on Labor day, Monday, Sept. 1, 2003) |
| Weeks of Sept. 8 and Sept. 15, 2003                     | Chapters 4 and 5:<br>Newton's Law of Motion, Further application of Newton's Laws   |
| Week of Sept. 22, 2003                                  | Chapter 6:<br>Gravitation and Newton's Synthesis  |
| Monday, Sept. 29, 2003                                  | First Term Exam (Chap. 1 - 6)   |
| Oct. 1 & Weeks of Oct. 6 and Oct. 13, 2003              | Chapters 7 - 9:<br>Work and Energy, Energy Conservation, Linear Momentum and Collisions                                     |
| Weeks of Oct. 20 and Oct. 27, 2003                      | Chapters 10 - 12:<br>Rotational Motion about a fixed axis, General Rorational Motion, Static<br>equilibrium                 |
| Monday, Nov. 3, 2003                                    | Second Term Exam (Chap 7 - 12)  |
| Nov. 5 & Weeks of Nov. 10 and Nov. 17,<br>2003          | Chapters 13 - 16:<br>Fluids, Oscillations, Wave Motion and sound  |
| Weeks of Nov. 24 and Dec. 1, 2003                       | Chapters 17 - 20:<br>Thermodynamics, Theory of Gases, Thermodynamic Laws  |
| Monday, Dec. 8, 2003                                    | Third (Final) Term Exam (Chap. 13 - 20)   |
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| Catrgory  | Dates  |          |
| 2)<br>  | Dates<br>Monday, Sept. 1, 2003   |          |
| <b>Catrgory</b><br>Labor Day  | Dates  |          |
| <b>Catrgory</b><br>Labor Day<br>First Term Exam   | Dates<br>Monday, Sept. 1, 2003<br>Monday, Sept. 29, 2003                         |          |
| <b>Catrgory</b><br>Labor Day<br>First Term Exam<br>Second Term Exam   | Dates<br>Monday, Sept. 1, 2003<br>Monday, Sept. 29, 2003<br>Monday, Nov. 3, 2003 |          |

### **Evaluation Policy**

- Term Exams: 50%
  - Total of three term exams
  - Best two of the three will be chosen
  - Each will constitute 25% of the total
  - Missing an exam is not permissible unless pre-approved
    - No makeup test
- Lab score: 20%
- Homework: 15%
- Pop-quizzes: 15%
- Extra credits: 10% of the total
  - Random attendances
  - Strong participation in the class discussions
  - Other many opportunities
- Will be on sliding scale unless everyone does very well



### Homeworks

- Solving chapter problems is the only way to comprehend class material
- An electronic homework system has been setup for you
  - Details are in the material distributed today
  - <u>https://hw.utexas.edu/studentInstruction</u>
     <u>s.html</u>
  - Roster will close next Wednesday, Sept. 3
- Each homework carries the same weight
- Home work will constitute <u>15% of the total</u> → A good way of keeping your grades high

Monday, Aug. 25, 2003



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### Attendances and Class Style

- Attendances:
  - Will be taken randomly
  - Will be used for extra credit
- Class style:
  - Lectures will be on electronic media
    - The lecture notes will be posted **AFTER** each class
  - Will be mixed with traditional methods
  - Active participation through questions and discussions are <u>STRONGLY</u> encouraged → Extra credit....



# Why do Physics?

- Exp. To understand nature through experimental observations and measurements (**Research**)
- Establish limited number of fundamental laws, usually with mathematical expressions
   Predict the nature's course

  - ? Theory and Experiment work hand-in-hand
  - ? Theory works generally under restricted conditions
  - ? Discrepancies between experimental measurements and theory are good for improvements
  - ? Improves our everyday lives, though some laws can take a while till we see amongst us



## What do we want from this class?

- Physics is everywhere around you.
- Understand the fundamental principles that surrounds you in everyday lives...
- Identify what law of physics applies to what phenomena...
- Understand the impact of such physical laws
- Learn how to research and analyze what you observe.
- Learn how to express observations and measurements in mathematical language.
- Learn how to express your research in systematic manner in writing
- I don't want you to be scared of PHYSICS!!!



# Brief History of Physics

- AD 18<sup>th</sup> century:
  - Newton's Classical Mechanics: A theory of mechanics based on observations and measurements
- AD 19<sup>th</sup> Century:
  - Electricity, Magnetism, and Thermodynamics
- Late AD 19<sup>th</sup> and early 20<sup>th</sup> century (Modern Physics Era)
  - Einstein's theory of relativity: Generalized theory of space, time, and energy (mechanics)
  - Quantum Mechanics: Theory of atomic phenomena
- Physics has come very far, very fast, and is still progressing, yet we've got a long way to go
  - What is matter made of?
  - How do matters get mass?
  - How and why do matters interact with each other?
  - How is universe created?



## Uncertainties

- Physical measurements have limited precision, however good it is, due to:
- Stat.{ Number of measurements
- Quality of instruments (meter stick vs micro-meter)
  Syst. Experience of the person doing measurements
  - In many cases, uncertainties are more important and difficult to estimate than the central (or mean) values



## Significant Figures

- Significant figures denote the precision of the measured values
  - Significant figures: non-zero numbers or zeros that are not place-holders
    - 34 has two significant digits, 34.2 has 3, 0.001 has one because the 0's before 1 are place holders, 34.100 has 5, because the 0's after 1 indicates that the numbers in these digits are indeed 0's.
    - When there are many 0's, use scientific notation:
      - $31400000 = 3.14 \times 10^{7}$
      - 0.00012=1.2x10<sup>-4</sup>



## Significant Figures

- Operational rules:
  - Addition or subtraction: Keep the <u>smallest number of</u> <u>decimal place</u> in the result, independent of the number of significant digits: 34.001+120.1=154.1
  - Multiplication or Division: Keep the <u>smallest</u>
     <u>significant figures</u> in the result: 34.001x120.1 = 4083, because the smallest significant figures is 4.



## Needs for Standards and Units

- Basic quantities for physical measurements
  - Length, Mass, and Time
- Need a language that everyone can understand each other
  - Consistency is crucial for physical measurements
  - The same quantity measured by one must be comprehendible and reproducible by others
  - Practical matters contribute
- A system of unit called <u>SI</u> (System International) established in 1960
  - Length in meters (*m*)
  - Mass in kilo-grams (kg)
  - Time in seconds (s)



## Definition of Base Units

| SI Units                        | Definitions  |
|---------------------------------|--|
| 1 <i>m (Length) =</i><br>100 cm | One meter is the length of the path traveled by light in vacuum during a time interval of <u>1/299,792,458 of a second</u> .   |
| 1 kg (Mass) =<br>1000 g         | It is equal to the mass of the international prototype<br>of the kilogram, made of platinum-iridium in<br>International Bureau of Weights and Measure in<br>France.  |
| 1 s (Time)                      | One second is the <u>duration of 9,192,631,770</u><br><u>periods of the radiation</u> corresponding to the<br>transition between the two hyperfine levels of the<br>ground state of the Cesium 133 (C <sup>133</sup> ) atom. |

There are prefixes that scales the units larger or smaller for convenience (see pg. 7)
Units for other quantities, such as Kelvins for temperature, for easiness of use



## International Standard Institutes

- International Bureau of Weights and Measure
   <u>http://www.bipm.fr/</u>
  - Base unit definitions:
  - http://www.bipm.fr/enus/3\_SI/base\_units.html
  - Unit Conversions: <u>http://www.bipm.fr/enus/3\_SI/</u>
- US National Institute of Standards and Technology (NIST) <u>http://www.nist.gov/</u>



#### Examples 1.3 & 1.4

Ex 1.3: A silicon chip has an area of 1.25in<sup>2</sup>. Express this in cm<sup>2</sup>.

.5 & 1.4  
1.25 in<sup>2</sup> = 1.25 in<sup>2</sup> × 
$$\left(\frac{2.54 \text{ cm}}{1 \text{ in}}\right)^2$$
  
= 1.25 in<sup>2</sup> ×  $\left(\frac{6.45 \text{ cm}^2}{1 \text{ in}^2}\right)$   
= 1.25 × 6.45 cm<sup>2</sup> = 8.06 cm<sup>2</sup>

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• Ex 1.4: Where the posted speed limit is 65 miles per hour (mi/h or mph), what is this speed (a) in meters per second (m/s) and (b) kilometers per hour (km/h)?

$$1 \text{ mi} = (5280 \text{ ft}) \left(\frac{12 \text{ in}}{1 \text{ ft}}\right) \left(\frac{2.54 \text{ cm}}{1 \text{ in}}\right) \left(\frac{1 \text{ m}}{100 \text{ cm}}\right) = 1609 \text{ m} = 1.609 \text{ km}$$
  
(a)  $65 \text{ mi/h} = (65 \text{ mi}) \left(\frac{1609 \text{ m}}{1 \text{ mi}}\right) \left(\frac{1}{1 \text{ h}}\right) \left(\frac{1 \text{ h}}{3600 \text{ s}}\right) = 29.1 \text{ m/s}$   
(b)  $65 \text{ mi/h} = (65 \text{ mi}) \left(\frac{1.609 \text{ km}}{1 \text{ mi}}\right) \left(\frac{1}{1 \text{ h}}\right) = 104 \text{ km/h}$   
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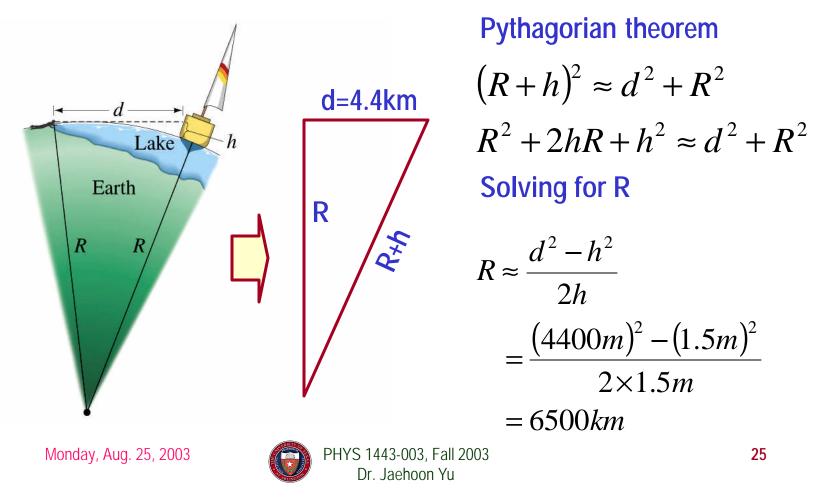
#### Estimates & Order-of-Magnitude Calculations

- Estimate = Approximation
  - Useful for rough calculations to determine the necessity of higher precision
  - Usually done under certain assumptions
  - Might require modification of assumptions, if higher precision is necessary
- Order of magnitude estimate: Estimates done to the precision of 10s or exponents of 10s;
  - Three orders of magnitude:  $10^3 = 1,000$
  - Round up for Order of magnitude estimate;  $8x10^7 \sim 10^8$
  - Similar terms: "Ball-park-figures", "guesstimates", etc



#### Examples 1.8

Estimate radius of the Earth using triangulation as shown in the picture when d=4.4km and h=1.5m.



### Dimension and Dimensional Analysis

- An extremely useful concept in solving physical problems
- Good to write physical laws in mathematical expressions
- No matter what units are used the base quantities are the same
  - Length (distance) is length whether meter or inch is used to express the size: Usually denoted as [L]
  - The same is true for *Mass ([M])* and *Time ([T])*
  - One can say "Dimension of Length, Mass or Time"
  - Dimensions are used as algebraic quantities: Can perform algebraic operations, addition, subtraction, multiplication or division



### Dimension and Dimensional Analysis

- One can use dimensions only to check the validity of one's expression: Dimensional analysis
  - Eg: Speed  $[v] = [L]/[T] = [L][T^{-1}]$ 
    - Distance (L) traveled by a car running at the speed V in time T
    - $L = V^*T = [L/T]^*[T] = [L]$
- More general expression of dimensional analysis is using exponents: eg.  $[v] = [L^n T^m] = [L] \{T^{-1}\}$ where n = 1 and m = -1



#### Examples

- Show that the expression [v] = [at] is dimensionally correct
  - Speed: *[v]* =L/T
  - Acceleration: [a] =L/T<sup>2</sup>
  - Thus, [at] = (L/T<sup>2</sup>)xT=LT<sup>(-2+1)</sup> =LT<sup>-1</sup> =L/T= [V]
- Suppose the acceleration *a* of a circularly moving particle with speed *v* and radius *r* is proportional to *r<sup>n</sup>* and *v<sup>m</sup>*. What are *n* and *m*?

